A HELIANGOLIDE FROM CHROMOLAENA GLABERRIMA

AHMED A. AHMED,* ALAN T. WHITTEMORE and TOM J. MABRY

Department of Botany, University of Texas at Austin, Austin, TX 78712 USA.

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Abstract—The investigation of *Chromolaena glaberrima* afforded, besides the known heliangolide chromolaenide, a new 4'-dehydrochromolaenide. The structure of the new compound was elucidated by spectroscopic methods.

INTRODUCTION

As part of our continuing chemosystematic investigations of sesquiterpene lactones of the Compositae, we investigated Chromolaena glaberrima (DC.) King and H. Robins., a member of the subtribe Praxelinae of the tribe Eupatorieae [1]. Previous workers have reported the heliangolide chromolaenide from this species [2], the only germacrane-derived sesquiterpene lactone reported so far from the subtribe [3] We report herein a similar heliangolide (2) with a novel side chain.

RESULTS AND DISCUSSION

The major sesquiterpene lactone isolated from Chromolaena glaberrima was chromolaenide (1), which was previously reported [2, 4] from the same species ¹³C NMR data, which had not been reported for 1, is listed in Table 1.

We report here a new heliangolide, namely, 4'-dehydrochromolaenide (2). High-resolution mass measurement of the molecular ion $(m/z \ 402)$ established the molecular formula of $2 \text{ as } C_{22}H_{26}O_7$. The presence of an aldehyde function in 2 was obvious from the IR (1690 cm⁻¹), ¹³C NMR (aldehydic carbon signal at δ 191.1) and ¹H NMR (aldehydic proton signal at δ 10.17) spectral data The ¹H NMR spectrum of 2 (see Table 2) was essentially identical to that of 1 except that the two-proton signal at δ 4.35 was replaced by the aldehydic proton signal at δ 10.17 (d) and the methyl group of the side chain shifted downfield from δ 1.8 to 2.3. The assignment of stereochemistry is based on the near identity of most of the NMR signals of 1 and 2 and on the co-occurrence of the compounds in C glaberrima.

Spin decoupling experiments were undertaken to confirm the structure of 2. Irradiation at δ 10.17, the frequency of the aldehydic proton, collapsed a multiplet at 6.75 (H-3') to a narrow quartet. Reciprocally, irradiation at δ 6.75 changed the doublet of the aldehydic proton at 10.17 and the signal for the methyl at 2.3 to singlets. In addition, irradiation at 2.3 simplified the multiplet at δ 6.75 to a clear

*Permanent address. Department of Chemistry, Faculty of Science, El-Minia University, El-Minia, Egypt

doublet, supporting coupling between the aldehydic and vinylic protons.

Since the aldehydic proton appeared at δ 10.17 and the side chain methyl group at 2.3, the stereochemistry of the C-2'/C-3' double bond was assigned the *E*-configuration. This is in agreement with previous observations that the chemical shift of aldehydic protons in trans- α , β -unsatu-

Table 1 ¹³C NMR spectral data for compounds 1 and 2*

	1	2	
C-1	125.3 d†	125.5 d†	
C-2	29 5 t	29 6 t	
C-3	769d‡	76 8 d‡	
C-4	137 7	1373	
C-5	126 5 d†	126 4 d†	
C-6	78.7 d‡	78.8 d‡	
C-7	48.5 d	48.4 d	
C-8	76 9 d‡	75 6 d‡	
C-9	43 4 t	43.3 t	
C-10	1366	1349	
C-11	135.6	1343	
C-12	170.1	169.7	
C-13	124.7 t	1249 t	
C-14	19 4 <i>q</i> †	196q†	
C-15	230q	23 1 q	
C-1'	166 3	1660	
C-2'	127.4	1268	
C-3'	142.7 d	144.4 d	
C-4'	59 6 t	191 1 d	
C-5'	12 5 q	119 <i>q</i>	
C-1"	1699	169.5	
C-2"	212q	212q	

^{*}Run at 22.6 MHz in CDCl₃ with TMS as internal standard. Assignments were made using off-resonance decoupling experiments and model compounds [7, 8]

^{†,‡}These assignments may be interchanged.

Table 2 1H NMR spectral data for compounds 1 and 2*

	1	2
H-1	5.27 m	5 27 m
H-2a	2.69 m	2.69 m
H-2b	2.29 m	2 29 m
H-3	5 27 t	5.27 t
H-5	$5\ 22\ dq\ (J=12,1.3)$	5.22 dq (J = 12, 1.3)
H-6	596dd(J=12,1.8)	5.96 dd (J = 12, 1.8)
H-7	2 98 m	2.98 m
H-8	5.27 m	5 27 m
H-9a	2.76 m	276 m
H-9b	2.45 m	2.45 m
H-13a	6.37 d (J = 2.2)	637d (J = 2.2)
H-13b	5.78 d (J = 20)	5.78 d (J = 2.0)
H-14	1.8 d	1.8 d
H-15	1.82d (J = 1.8)	182d(J=18)
H-3'	6.8 m	675m (J = 12, 2)
H-4'	4.35 m	$1017d\ (J=13)$
H-5'	1.8 d (J = 1 3)	23d(J=2)

*Run at 200 MHz in CDCl₃ with TMS as internal standard. Coupling constants in Hz

rated systems (E-form) generally appear at lower field (≥ 10 ppm) than the same proton in $cis-\alpha,\beta$ -unsaturated systems, which usually appear at 9.26–9.4 [5]. This is the first report of this side chain.

EXPERIMENTAL

Extraction of Chromolaena glaberrima. Leaves (500 g) were collected by Alan Whittemore on 18 March 1982, between Tepic and Jalcocotan, Nayarit, Mexico (voucher in Herb TEX). The unground plant material was washed with CH₂Cl₂ and the extract worked up in the usual manner [6] The crude syrup obtained (75 g) was chromatographed on a silica gel column (160 g) packed in CH₂Cl₂. The column was eluted with a CH₂Cl₂-*i*-PrOH gradient, increasing the polarity with *i*-PrOH; 20 fractions of 300 ml each were collected. Fractions 7-11 (100% CH₂Cl₂) gave 500 mg 1 Fractions 13-16 (1% *i*-PrOH) gave compound 2 with a trace of 1. Compound 2 was separated by prep. TLC (silica gel, 15 mm, CH₂Cl₂-*i*-PrOH, 15.1) to give 30 mg of an oily material.

Chromolaenide (1) This compound was previously isolated from Chromolaena glaberrima [2]. All the spectral data (MS, IR and ¹H NMR) and mp are consistent with the former report.

4'-Dehydrochromolaenide (2) MS (probe) 70 eV, m/z (rel. int.): 402 [M]⁺ (4), 289 [M – OHC–CH=C(Me)CO₂]⁺ (84), 288 [M – OHC–CH=C(Me)COOH]⁺ (9 5), 228 [288 – HOAc]⁺ (100), 97 [OHC–CH=C(Me)CO]⁺ (91) IR $v_{\rm max}^{\rm CHCl_3}$ cm⁻¹ 2950 (aldehyde C–H), 1760 (lactone C=O), 1690 (aldehyde C=O), UV $\lambda_{\rm max}^{\rm MeOH}$ nm (log ε): 213 (2200).

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